

NAME \_\_\_\_\_

ID # \_\_\_\_\_

ORGANIC CHEMISTRY I (2301-003)

1:25 – 2:15 pm, October 22, 2007

Exam 2

When the exam begins, write your name at the top of each page.

You may use pen or pencil. However, re-grades will be considered only for exams completed in pen.

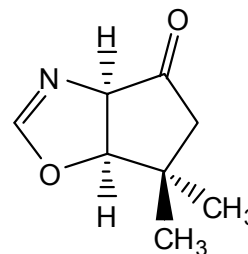
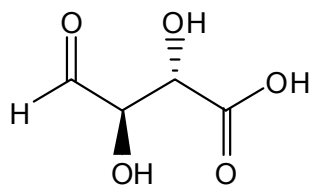
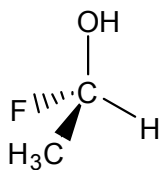
Please write your answers in the boxes/spaces provided. If your answer is not in the appropriate space (say, for example, it's on the back of the page), draw us an arrow and/or note telling us where to look.

NAME \_\_\_\_\_

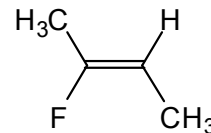
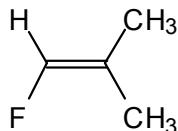
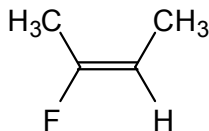
Scoring: 1. \_\_\_\_\_ / 10      5. \_\_\_\_\_ / 9  
2. \_\_\_\_\_ / 6      6. \_\_\_\_\_ / 27  
3. \_\_\_\_\_ / 8      7. \_\_\_\_\_ / 20  
4. \_\_\_\_\_ / 20

Total Score: \_\_\_\_\_ / 100

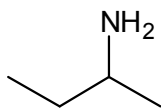
1. (10 pts) On the structures below, label each chiral center with its appropriate Cahn-Ingold-Prelog designation [(*R*) or (*S*)].



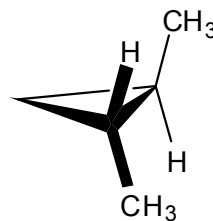
2. (6 pts) Label the stereochemistry at each olefin below as (*E*), (*Z*), or “neither”.



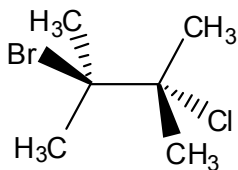
3. (8 pts) Are the following molecules chiral or achiral? For each structure, **circle one answer**.



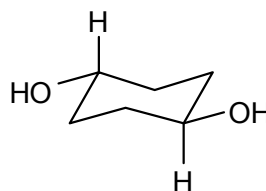
**CHIRAL** or **ACHIRAL** ?



**CHIRAL** or **ACHIRAL** ?

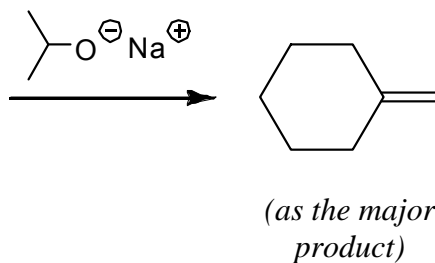
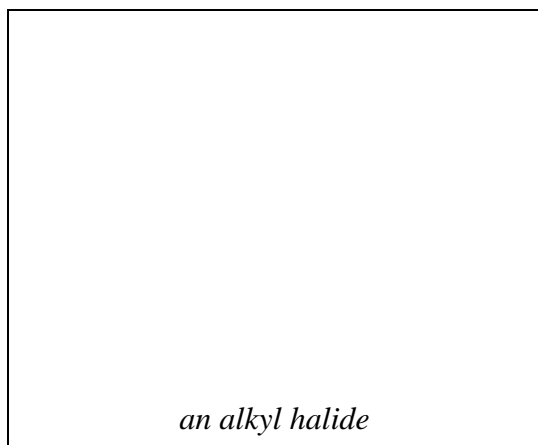


**CHIRAL** or **ACHIRAL** ?

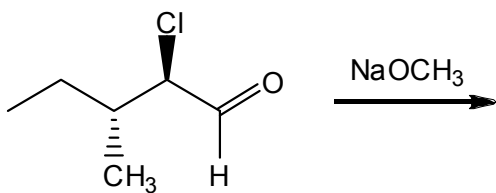
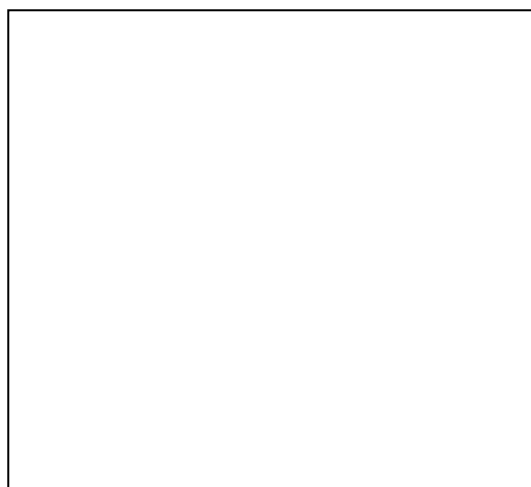
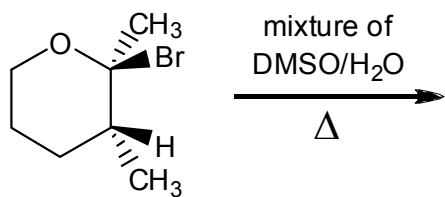
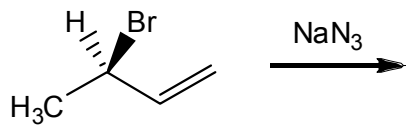


**CHIRAL** or **ACHIRAL** ?

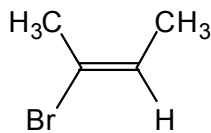
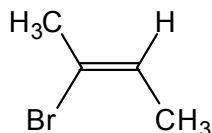
4. (20 pts) Draw the missing reactant or product in the empty boxes. For products, give the predominant, most favored product.



*(Problem 4 continues on next page)*



5. (9 pts) How would you describe the relationship between each of the pairs of structures below? Are they enantiomers or diastereomers, or are they just two ways of illustrating the same molecule? **Circle one answer** for each pair.



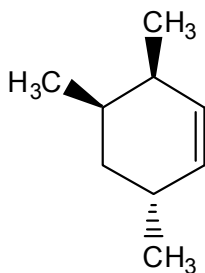
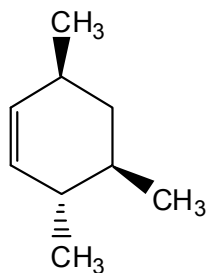
**ENANTIOMERS**

*or*

**DIASTEREOMERS**

*or*

**SAME MOLECULE**



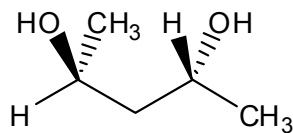
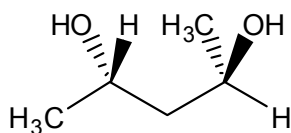
**ENANTIOMERS**

*or*

**DIASTEREOMERS**

*or*

**SAME MOLECULE**



**ENANTIOMERS**

*or*

**DIASTEREOMERS**

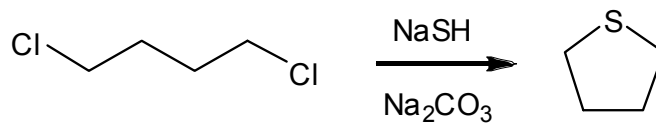
*or*

**SAME MOLECULE**

6. (27 pts) For each of the reactions shown below, draw a mechanism that explains how each product is generated from the starting material. In your answers, make sure that you:

- Draw each step of the mechanism separately;
- Use “electron pushing” to show where the electrons in each step go.

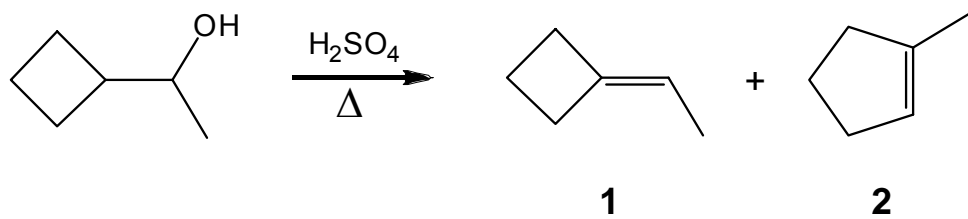
(a)



A large empty rectangular box provided for drawing the reaction mechanism.

*(Problem 6 continues on next page)*

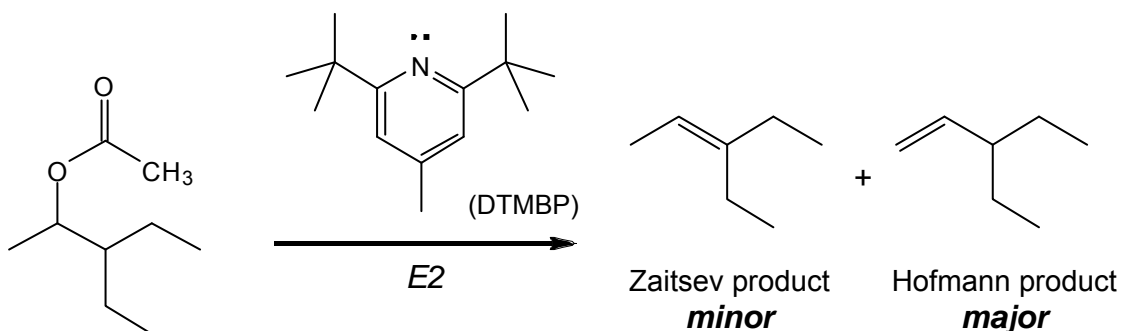
(b)



*mechanism for product 1*

*mechanism for product 2*

7. (20 pts) When the starting material shown below is combined with 2,6-di-tert-butyl-4-methylpyridine (DTBMP), a sterically hindered base, the resulting E2 elimination predominantly yields the *less* substituted (Hofmann) alkene product.



In the space below, explain this preference using a potential energy diagram that illustrates:

- Two potential energy curves, one for each product.
- In the brackets, the structure of the rate-determining transition state for each pathway.
- The relative potential energies of all starting materials, transition states, intermediates (if any), and products. Show which energies correspond to which chemical structures.
- Two overall activation energies,  $E_a[\text{Zaitsev}]$  and  $E_a[\text{Hofmann}]$ , for the two pathways.

